



## Porous coordination polymers (PCPs) based on redox ligands

Submitted by Marie-Claire Dul on Fri, 06/12/2015 - 09:50

Titre	Porous coordination polymers (PCPs) based on redox ligands
Type de publication	Communication
Type	Communication par affiche dans un congrès
Année	2015
Langue	Anglais
Date du colloque	18-20/05/2015
Titre du colloque	Journées Scientifiques 2015 de la Société chimique de France, Section Bretagne-Pays de la Loire
Auteur	Leroux, Maxime [1], Dul, Marie-Claire [2], Toma, Oksana [3], Allain, Magali [4], Mercier, Nicolas [5]
Pays	France
Ville	Nantes
Mots-clés	bipyridinium [6], gas storage [7], molecular recognition [8], porous coordination polymers [9], Redox active ligands [10]
Résumé en anglais	<p>PCPs (Porous Coordination Polymers) or MOFs (Metal Organic Frameworks) are crystalline materials whose structures consist of metal-based nodes bridged by organic linking groups. They are a well known class of porous materials which can have applications in gas storage (H<sub>2</sub>, CH<sub>4</sub>, CO<sub>2</sub>,...), heterogeneous catalysis and chemical sensors.<sup>1</sup> Up to now the main strategy to increase the absorption properties has been to introduce coordinatively unsaturated metal centres. In contrast, the incorporation of cationic organic ligands is much rarer despite a certain potential as shown by some reports.<sup>2</sup> Our original approach consists of mixing electro-active viologen derivatives to a useful coordination function like carboxylate, which is widely used in such materials,<sup>3</sup> to synthesize new PCPs for gas storage and molecular recognition. Those ligands are based on N-substituted-4,4'-bipyridinium monocation and N,N'-disubstituted-4,4'-bipyridinium dication carrying one or several carboxylate groups. Our synthetic strategy and new results will be described in this poster, taking the [Cd<sub>4</sub>Cl<sub>6</sub>L<sub>3</sub>](CdCl<sub>4</sub>) compound as example, with L = 4,4'-bipy-(C<sub>6</sub>H<sub>4</sub>COO)<sub>2</sub>. In addition to provide a highly stable structure upon temperature and moisture, this PCP exhibits accessible channels with a large zwitterionic surface area which allow reversible sorption properties of gas and small molecules. The obvious color shift in presence of ammonia vapors offers a high potential for chemical sensors and optical applications.</p>
URL de la notice	<a href="http://okina.univ-angers.fr/publications/ua12472">http://okina.univ-angers.fr/publications/ua12472</a> [11]

### Liens

- [1] [http://okina.univ-angers.fr/publications?f\[author\]=21730](http://okina.univ-angers.fr/publications?f[author]=21730)
- [2] <http://okina.univ-angers.fr/m.dul/publications>
- [3] [http://okina.univ-angers.fr/publications?f\[author\]=19068](http://okina.univ-angers.fr/publications?f[author]=19068)
- [4] <http://okina.univ-angers.fr/magali.allain/publications>

- [5] <http://okina.univ-angers.fr/nicolas.mercier/publications>
- [6] [http://okina.univ-angers.fr/publications?f\[keyword\]=17008](http://okina.univ-angers.fr/publications?f[keyword]=17008)
- [7] [http://okina.univ-angers.fr/publications?f\[keyword\]=18531](http://okina.univ-angers.fr/publications?f[keyword]=18531)
- [8] [http://okina.univ-angers.fr/publications?f\[keyword\]=17007](http://okina.univ-angers.fr/publications?f[keyword]=17007)
- [9] [http://okina.univ-angers.fr/publications?f\[keyword\]=17006](http://okina.univ-angers.fr/publications?f[keyword]=17006)
- [10] [http://okina.univ-angers.fr/publications?f\[keyword\]=5063](http://okina.univ-angers.fr/publications?f[keyword]=5063)
- [11] <http://okina.univ-angers.fr/publications/ua12472>

Publié sur *Okina* (<http://okina.univ-angers.fr>)